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INVESTIGATION OF THE ADSORPTION OF NONPOLAR ADSORBATE MOLECULES ON THE ILLITE SURFACE

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Краткое описание

The full thermodynamic characteristics of CO₂ and CH₃OH adsorption on the mineral illite were obtained using the adsorption - calorimetric method. The crystal chemistry of illite was studied using the polar CH₃OH molecule and the quadrupole CO₂ molecule as a molecular probe. The mechanism of adsorption of these molecules, the conformation of the resulting ion/molecular complexes, and the thermokinetics of adsorption were revealed. The adsorption isotherms are studied and described by their corresponding equations. The molecular mechanism of CO₂ and CH₃OH adsorption on illite was studied in the entire filling area. It has been established that under the influence of a polar methanol molecule, K⁺ migrate to the basal and lateral faces of illite, filling vacant holes with six-dimensional CH₃OH/K⁺ complexes of illite, the state of molecules on the surface is close to liquid - like. The process of migration of cations is completely reversible. It was found that the adsorption of a quadrupole CO₂ molecule on illite does not lead to the migration of cations. Adsorption proceeds on K⁺ cations, which occupy the basal one third of the illite surface. © 2023 Journal of Chemical Technology and Metallurgy. All Rights Reserved.

Ключевые слова автора

adsorbate; adsorption; carbon dioxide; illite; isotherm; methanol; thermodynamics

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